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# Chemical Analysis of Microplastics in Sachet Water Samples in Katsina Metropolis: Improving Health and Safety Standards

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#### **ABSTRACT**

The assessment of microplastics (MPs) in sachet water samples sold in Katsina metropolis was investigated. The absorption peak of the infrared spectroscopy (IR) obtained were 1640, 2117 and 3265 cm<sup>-1</sup> for sample S<sub>A</sub>; 1192, 1640, 2124, 3265 cm<sup>-1</sup> for  $S_B$ ; for sample  $S_C$ : 2035, 2146, 2236, 2333, 3227, 3667, 3734, 3816 cm<sup>-1</sup>,  $S_D$ : 1640, 2117, 3265 cm<sup>-1</sup>, sample S<sub>E</sub>: 1640, 2124, 3265 cm<sup>-1</sup> and S<sub>F</sub>: 1640, 2117, 3265 cm<sup>-1</sup>. The absorption peak of samples S<sub>A</sub> to S<sub>F</sub> that correspond to functional groups were C=C, C=O, O-H, except S<sub>C</sub> that have additional C-O. Comparatively, all the studied samples were absorbed at 1640 cm<sup>-1</sup> while other absorption peaks varied. The particle size of the sample ranged from 32.0 particle/0.75L sample S<sub>A</sub> to 64.1 particle/0.75L sample S<sub>F</sub>. The microplastic detected were generally polyethylene terephthalate (PET), polyethylene (PE), and polyvinyl chloride (PVC) and granules in shape. Microplastic pollution load index (MPPLI), micro-plastic contamination factor (MPCF), and the estimated daily intake (EDI) for both adults and children were determined. The MPCF values obtained ranged from 1.00 for sample S<sub>A</sub> to 2.00 for sample S<sub>C</sub> with a series of profile as S<sub>A</sub>< S<sub>B</sub><S<sub>F</sub><S<sub>E</sub><S<sub>D</sub><S<sub>C</sub>. The result for EDI for adults was slightly greater than 1 except in all the samples except sample S<sub>C</sub> which was 2.01 depicting moderate and high daily intake in adults respectively. The EDI for children ranged from 3.84 in sample S<sub>A</sub> to 7.69 in sample S<sub>C</sub>. The EDI result showed that children consume more microplastics than adults. The MPPLI obtained was 1.28. Microplastic in sachet water require significant attention to reduce the menace it may cause in the health status of humans due to high daily intake studied.

**Keywords:** Microplastic, Sachet water, Health risk, Polymer, Contamination

#### INTRODUCTION

Microplastics are classified as synthetic plastic fragments that are often invisible to the naked eye but have significant environmental and health implications. Their presence in drinking water stems from various sources, including the degradation of plastic packaging, industrial emissions, and atmospheric deposition [11]. These particles can harbor toxic chemicals and microorganisms, increasing health risks when consumed [4]. Microplastic contamination has become a global environmental and health concern due to the widespread use of plastic materials in modern society. Andrady, [3] see microplastics as plastic particles smaller than 5 mm, that originate from the breakdown of larger plastic debris (secondary microplastics) or are directly manufactured as microbeads or pellets (primary microplastics). These particles find their way into terrestrial and aquatic environments through improper waste management and industrial activities, eventually infiltrating drinking water sources.

Globally, the issue of microplastic contamination in drinking water has gained significant attention due to its potential health risks. Mason *et al.*, [8] identified synthetic polymer contamination in bottled water, attributing it to mechanical stress and the leaching of microplastics from packaging materials. Terefe et al. [12] documented microplastic particles in table salts, emphasizing the widespread nature of plastic pollution and its infiltration into everyday consumables.





The ubiquitous presence of plastics in ecosystems has led to microplastics infiltrating water systems, food chains, and human bodies. In regions like Katsina Metropolis, where sachet water is a primary source of drinking water, concerns about microplastic contamination are exacerbated by poor waste disposal practices, substandard

packaging, and harsh storage conditions such as prolonged exposure to ultraviolet (UV) radiation Okpashi *et al.*, [10]. Despite the known risks of microplastics, ranging from their potential to act as vectors for toxic pollutants to causing cellular damage when ingested, there is limited empirical data on their prevalence in sachet water in Katsina. The absence of such data undermines efforts to establish effective regulations, improve packaging standards, and educate the public on safe water handling practices. Without intervention, the health risks associated with microplastic ingestion may become an unaddressed public health crisis in the region. While existing studies have focused on neighboring regions, this research will provide localized insights into Katsina Metropolis.

Sachet water, a widely consumed and affordable drinking water option in Katsina particularly and Nigeria at large, is not exempt from this issue. Studies suggest that sachet water is particularly prone to contamination due to poor storage conditions, leaching of packaging materials, and environmental exposure (Aliyu et al., [1].

Similarly, Udoh *et al.*, [13] explore the issue of water inequality in Chiapas, Mexico, where despite abundant water resources, over half of the population, particularly in rural and indigenous areas, lacks access to basic water services. This disparity contributes to health issues, including waterborne diseases. The study calls for a transdisciplinary approach to water management that addresses social inequalities and promotes sustainable, community-based solutions to improve water access and public health.

In Africa, Bazaanah and Mothapo (2023) conducted a comprehensive analysis of water and sanitation systems in the rural communities of Lepelle Nkumpi Local Municipality, South Africa. Utilizing a mixed-methods approach, they gathered data from 657 household and institutional respondents. The study revealed that households use water for various purposes, including consumption, domestic chores, and productive activities.

In Nigeria, sachet water serves as an affordable and widely consumed alternative to other drinking water sources. However, it is not exempt from contamination risks. Alivu et al., [1] reported significant microplastic pollution in sachet water from Kaduna Metropolis, identifying poor packaging standards, environmental degradation, and improper storage conditions as major contributors. Okpashi et al., [10] revealed that exposure to ultraviolet (UV) rays during storage accelerates the degradation of plastic packaging, releasing microplastics into the water. These studies highlight a systemic challenge, pointing to inadequate regulations and monitoring in the sachet water industry across Nigeria. Kusa & Joshua [7] study assessed the physicochemical and biological properties of various sachet water brands in Nigeria. Findings indicated that while most parameters met standard guidelines, certain brands exhibited elevated levels of lead and iron. Notably, all sampled brands were contaminated with Escherichia coli, posing significant health risks. The study underscores the necessity for stringent quality control measures and regular monitoring to ensure the safety of sachet water consumed by the public. Udoh et al., [13] highlighted concerns regarding the microbial quality of packaged drinking water, including sachet water, in Nigeria. The study revealed that some products did not comply with World Health Organization (WHO) and Nigerian Industrial Standards (NIS) for drinking water quality, detecting pathogens such as Escherichia coli and Salmonella species. These findings emphasize the critical need for improved regulatory oversight and adherence to safety standards to protect public health.

However, Imam *et al.*, [5] assesses the state of drinking water quality monitoring in Northern Nigeria, revealing that only 13.11% of the population has access to clean water. Additionally, 31.14% of water sources were deemed 'fair,' necessitating further treatment to prevent health issues due to contamination levels not meeting WHO standards. The study emphasizes the urgent need for comprehensive monitoring and intervention to meet Sustainable Development Goal 6 by 2030. Adesakin *et al.*, [2] evaluates the physicochemical and microbial quality of sachet water brands in the Samaru community, Zaria. Findings indicated that while most physicochemical parameters met WHO [15] standards, there were concerns regarding microbial contamination due to prolonged storage and inadequate handling practices. The study





analyzed the different types of Microplastic in sachet water samples and assessed the risk exposure of humans to the existing Microplastic based on daily intake in both adults and children of the sampled sachet water of Katsina metropolis.

#### **Materials and Methods**

#### Study Area

The study was carried out in the Katsina metropolis, Katsina State. It is located in the North-Western part of Nigeria at: 12° 59'7.9116 N and 7° 37'1.7184"E.

#### **Sample Collection**

Six different brands of sachet water samples from different sampling sites were purchased in triplicate in Katsina Metroplis, Katsina State, Nigeria and labeled  $S_A$ ,  $S_B$ ,  $S_C$ ,  $S_D$ ,  $S_E$ , and  $S_F$ . They were stored in an ice chest and transported to the laboratory for analysis.

#### **Sample Processing**

The water samples were filtered using Whatman 1823-047 grade GF/D glass fibre filter paper with a pore size of 2.7 µm in other to separate the MNPs from the bulk water. The MNPs obtained from filtration were thoroughly washed with distilled water and oven-dried at 65 °C. The mass of the MNPs was then determined by weighing the dried solid MNPs using an analytical balance to the nearest 0.1 mg.

#### MPs Characterization

Visual Sorting – This identifies MNP numbers, sizes, shapes and colours. MNPs were identified using a Nikon SMZ 745T stereomicroscope at 20–40× magnification [9].

#### **Procedure for FTIR**

Buck scientific M530 USA FTIR was used for the analysis. This instrument was equipped with a detector of deuterated triglycinesulphate and beam splitter of potassium bromide. The software of Gram  $S_A$  was used to obtain the spectra and to manipulate them. An approximately of 1.0g of samples, 0.5ml of nujol was added, they were mixed properly and placed on a salt pellet. During measurement, FTIR spectra were obtained at frequency regions of 4,000-600 cm-1 and co- added at 32 scans and at 4 cm-1 resolution. FTIR spectra were displayed as transmitter values.

#### Health Risk Assessment of Microplastics: Microplastics contamination factors and pollution load index

The microplastics contamination factors (MPCfs) and pollution load index (MPPLI) in the sachet water were measured as described in previous studies Verla *et al.*, [13]. The MPCf refers to the contamination of MPs in the studied drinking water (Sachet water) compared to the background values. The MPCf and MPPLI were mathematically computed using equations (1) and (2). Where MPi is the quantity of MPs in sample i while MPb is the minimum baseline concentration taken from the lowest MPs abundance recorded in the study of Mason *et al.*, [8] as it shares similar environments and analytical context as this study.



MPi = quantity of MPs in ith sample; MPb = the minimum baseline concentration taken from the lowest MPs abundance recorded in the study of Verla *et al.*, (2019) which has similar environment and analytical context as this study.

$$MPPLI = (MPCF1 \times MPCF2 \times MPCF3...MPCFn)^{1/n}...(2)$$

The MPCfs were categorized according to (Verla *et al.*, 2019). Values with MPCf < 1 are low contamination,  $1 \le MPCf < 3$  are moderately contaminated,  $3 \le MPCf \le 6$  are considerably contaminated and MPCf  $\ge 6$  very highly contaminated.





#### **Estimated Daily Intake:**

An individual risk pathway as a result of human exposure to microplastic contamination of drinking water could be through oral ingestion Aliyu *et al.*, [1]. Therefore, the estimated daily intake (EDI) due to exposure to overall MPs resulting from ingestion of contaminated water is determined using equation 3.

 $EDI = MPI \times RI/Bw...(3)$ 

EDI = estimated daily intake of MPs based on quantity through ingestion of the drinking water (particle/L/Bw-day)

MPi = average quantity of the MPs in drinking water (MP particle/L)

RI = ingestion rate (2.2 L/day for adults; 1.8 L/day for children)

Bw = average body weight (70 kg for adults; 15 kg for children)

#### RESULTS AND DISCUSSION

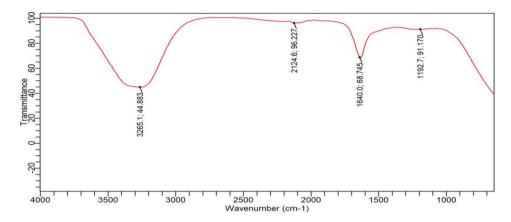
The Fourier Transform Infrared (FTIR) spectroscopy was conducted to ascertain the functional groups present in the sample under investigation [9]. The FTIR works on the principle that the molecule vibrates at specified frequencies and ranges from 200 cm<sup>-1</sup> to 4000 cm<sup>-1</sup> which falls within the IR portion of the electromagnetic spectrum. When an IR is incident on a sample, it absorbs radiation at a frequency similar to its molecular vibration frequency and transmits other frequencies. The infrared spectrum is obtained. The table below shows the results obtained from the infrared spectrum of the studied samples. From the FTIR results of the analysis carried out on the bottled water samples, the absorption peak of sample  $S_A$  were 1640, 2117 and 3265 cm<sup>-1</sup>;  $S_B$ : 1192, 1640, 2124, 3265;  $S_C$ : 2035, 2146, 2236, 2333, 3227, 3667, 3734, 3816;  $S_D$ : 1640, 2117, 3265;  $S_E$ : 1640, 2124, 3265 cm<sup>-1</sup> and  $S_F$ : 1640, 2117, 3265 cm<sup>-1</sup>. There were three absorption peaks each of samples  $S_A$ ,  $S_E$  and  $S_F$  with corresponding functional group C=C, C=O, C=O, C=O, C=O, C=O, C=O.

Samples	Polymer Type	Absorbance	Functional Group
		Peak (cm <sup>-1</sup> )	
S <sub>A</sub>	PE, PET	1640, 2117, 3265	C=C, C=O, O-H
$S_B$	PE, PET	1192, 1640, 2124, 3265	C=C, C=O, O-H
S <sub>C</sub>	PE, PET, PVC	2035, 2146, 2236, 2333, 3227, 3667, 3734, 3816	C-O, C=C, $C \equiv C$ , O-H
S <sub>D</sub>	PE, PET	1640, 2117, 3265	C=C, C=O, O-H
S <sub>E</sub>	PE, PET	1640, 2124, 3265	C=C, C=O, O-H
S <sub>F</sub>	PE, PET	1640, 2117, 3265	C=C, C=O, O-H

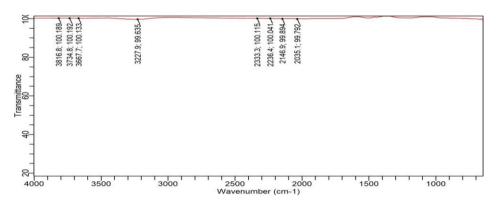
# **Table. 1** FTIR Results of the Samples S<sub>A</sub> to S<sub>F</sub>

Fig 1. FTIR Spectra in Sachet water samples of S<sub>A</sub>, S<sub>B</sub>, S<sub>C</sub>, S<sub>D</sub>, S<sub>E</sub> and S<sub>F</sub>.

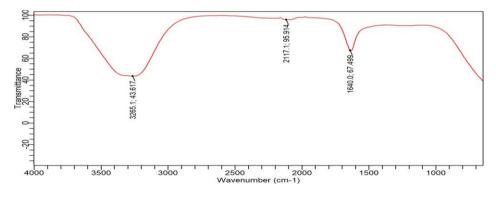
## $S_{\mathbf{A}}$



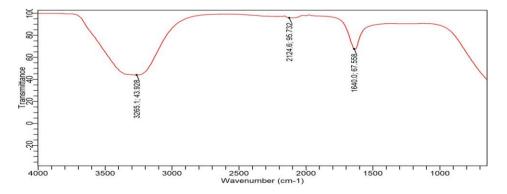
## $S_B$



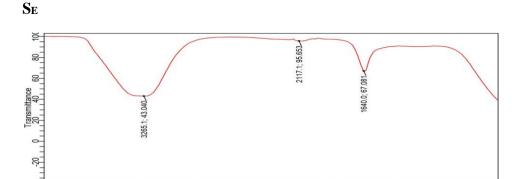
## $\mathbf{S}_{\mathbf{C}}$

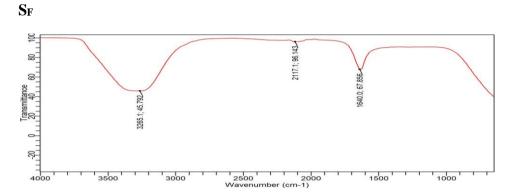


## $S_D$









The studied samples were all absorbed at 1640 cm<sup>-1</sup> while other absorption peaks varied. The IR result obtained by [11] in plastic bottled water showed strong peaks at 2916 cm<sup>-1</sup>, 2846 cm<sup>-1</sup>, and 2914 cm<sup>-1</sup> showing C-H stretch, CH<sub>2</sub> bend at the peak of 1466 cm<sup>-1</sup>, 1462 cm<sup>-1</sup> and 1747 cm<sup>-1</sup> indicating C=O, while 1241 cm<sup>-1</sup>, 1035 cm<sup>-1</sup> indicate C-O bond stretching. The results were applicable to what was obtained in sample S<sub>A</sub>, S<sub>B</sub>, S<sub>C</sub>, S<sub>D</sub> and S<sub>F</sub> of this study. Kusa and Joshua in their study obtained 2920 cm<sup>-1</sup>, and 2850 cm<sup>-1</sup> to show C-H bond stretching vibration [10]. The presence of a peak at 2035, 2146, 2236 cm<sup>-1</sup> in sample S<sub>C</sub> corresponding to C-O stretching vibration indicates the presence of ester groups. This suggests that the plastic identified in the sample may be composed of polyester materials which are commonly used in the production of various plastic products including packaging materials [11]. Furthermore, the detection of peaks at 2117 cm<sup>-1</sup>, 2035 cm<sup>-1</sup>, 2124 cm<sup>-1</sup> and in all sachet water samples is an indication of the presence of C=O bond stretching vibration pointing toward the presence of carbonyl groups in the polymer and the presence of 1192 cm<sup>-1</sup> and 1640 cm<sup>-1</sup> in sample S<sub>A</sub>, S<sub>D</sub> and S<sub>F</sub> and all other samples respectively represent C=C bond of an aromatic. These C- O and C=C arid maybe associated with plastic like polyethylene terephthalate (PET), mostly used materials for the manufacturing of beverage bottles and food containers. More so, the identification of peaks at 3265 cm<sup>-1</sup>, 3667 cm<sup>-1</sup>, 3734 cm<sup>-1</sup>, and 3816 cm<sup>-1</sup> in all Samples and mostly in S<sub>C</sub> correspond to O-H stretching vibration, suggesting the presence of a hydroxyl group in the sample. This O-H can be attributed to the additive or surface modification applied to the plastic materials. In the determination of micro-plastic in water samples in various other studies, the most frequently identified plastics were PE, PET, PP, PS, and PVC[1]. This is similar to the PE, PET, and PVC obtained in this study.

**Table 2**. Results of SEM characterization of MPs in the Bottled Water Samples

Samples	Brand Location	Microplastics Composition	Particle Size	Shape
S <sub>A</sub>		PE, PET, PVC	32.0	Granules
$S_B$		PE, PET, PVC	36.2	Granules
$S_{\rm C}$		PE, PET, PVC	64.1	Granules
$S_D$		PE, PET, PVC	41.4	Granules





S <sub>E</sub>	PE, PET, PVC	39.1	Granules
$S_{\mathrm{F}}$	PE, PET, PVC	38.5	Granules

#### Scanning Electron Microscope and Energy Dispersive X-ray Spectroscopy Result

Scanning Electron Microscope (SEM) is a technique used in obtaining the surface morphology of a substance by scanning the surface of the material to create a high-resolution image [9]. The resulting image gives detailed information about the object and its physical characteristics while Energy Dispersive X-ray Spectroscopy (EDS) gives information about the elemental composition of the substance under investigation[8-9]. The table 2 above shows the result of SEM obtained from the studied sachet water samples of S<sub>A</sub> to S<sub>F</sub>. The table showed the particle size of the sample ranged from 32.0 particle/0.75L sample S<sub>A</sub> to 64.1 particle/0.75L sample S<sub>F</sub>. The MPs types detected were polyethylene terephthalate (PET), polyethylene (PE), and polyvinyl chloride (PVC) while granules were the predominant shape of MPs obtained.

However, table 3 below shows the result of the health risk assessment of the microplastics in sachet water samples. The micro-plastic pollution load index (MPPLI), micro-plastic contamination factor (MPCF), and estimated of daily intake (EDI) were determined as well. The MPCF values ranged from 1.00 for sample  $S_A$  to 2.00 for sample  $S_C$  and followed the profile,  $S_A < S_B < S_C < S_C < S_C < S_C$ . The EDI values of the samples were evaluated for both adults and children. The EDI values of all samples for adults were slightly greater than 1 which is an indication of a moderate daily intake of MPs suggesting that daily consumption may be moderately risky except for sample  $S_C$  showed 2.00 EDI in adults which is high daily intake and may be high risk[11]

#### Health Risk Assessment of MPs in Sachet Water Samples

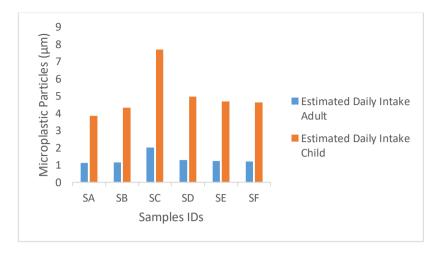
**Table 3.** Estimated Daily Intake of Microplastics in Sampled Sachet Water

Samples	MPCF	Estimated Daily Intake	
		Adult	Child
$S_A$	1.00	1.13	3.84
$S_{\mathrm{B}}$	1.13	1.14	4.34
$S_{C}$	2.00	2.01	7.69
$S_{\mathrm{D}}$	1.29	1.30	4.97
S <sub>E</sub>	1.22	1.23	4.69
$S_{\mathrm{F}}$	1.20	1.21	4.62
	MPLLI = 1.28		

However, the EDIs for children ranged from 3.84 in sample  $S_A$  to 7.69 in sample  $S_C$ . This is an indication that children consume more MPs than adults. This result is in agreement with the study of Kusa and Joshua [7], that the EDI for children is always higher compared to that of adults. From the results, it is obvious that children are more likely to consume MPs beyond the threshold level than adults. However, there is no backing evidence of the danger of MP consumption to human health. The MPPLI was calculated to be 1.27 which is lower than what was obtained by Aliyu *et al.*, [1]. Also, Figure 2 below showed that micro-plastic are more abundant in the sachet water samples in  $S_C$  and exposure is more significant among the consumers of the samples compared to other samples investigated. The lowest levels of MP in the sachet water samples was found in samples  $S_A$ .



Fig. 2. Daily Intake of Microplastics particles in Adult and Child



**Table 4.** Microplastic Contamination Factors and Pollution Load Index in Sachet Water

Samples	MPCF	Risk Category
S <sub>A</sub>	1.00	Moderately Contaminated
S <sub>B</sub>	1.13	Moderately Contaminated
S <sub>C</sub>	2.00	Moderately Contaminated
$S_D$	1.29	Moderately Contaminated
SE	1.22	Moderately Contaminated
S <sub>F</sub>	1.20	Moderately Contaminated

#### **CONCLUSION**

The result obtained showed the presence of polymers. The IR result from the analysis of the Sachet samples showed the presence of polymers like PE, PET, and PVC. The microplastic shape was granules from the surface morphology of the SEM results conducted. The EDI for Children was higher than the adults. This is an indication that children are more susceptible to Microplastic infiltration than adults. As water is life to both plants and animals it is detrimental and life threatening when polluted form of water is ingested. Regulatory monitoring, stricter packaging standards and penalties for non-compliance and should be adopted, also Public health campaigns could be proposed to educate consumers about safer storage practices, such as avoiding UV exposure. Finally future studies might integrate biological assays to examine the physiological impacts of detected Microplastic levels, strengthening the connection between environmental exposure and human health outcomes.

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#### **Conflict Of Interest**

There is no conflict of interest during the course of this research.

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